Unit IV

7. (a) Discuss the principle of electronic spectroscopy. Give its applications also.

5

(b) State and explain the rotational spectra of diatomic molecules. Discuss its application.

Or

- 8. (a) Illustrate the basic principle of fluorescence spectroscopy. Discuss the theory, instrumentation and important applications of it in medicine.
 - (b) Summarize diffraction and scattering techniques.

Compulsory Question

9. (a) State and explain the role of doping on band structures.

No. of Printed Pages: 5 Roll No.

18A10

B.Tech. EXAMINATION, 2024

(First Semester)

(C Scheme) (Re-appear Only)

CH101C

(Common for All Branches)

CHEMISTRY

Time: 3 Hours

[Maximum Marks: 75

Before answering the question-paper candidates should ensure that they have been supplied to correct and complete question-paper. No complaint, in this regard, will be entertained after the examination.

Note: Attempt *Five* questions in all, selecting *one* question from each Unit. Q. No. 9 is compulsory. All questions carry equal marks. Draw neat diagrams wherever applicable.

Unit I

- 1. (a) Illustrate solution of particle in one dimensional box. Discuss its application for conjugated molecule.
 - (b) Explain wave function and electron negativity in details.

Or

- 2. (a) Describe the π -molecular orbital of butadiene molecule.
 - (b) Define polarizability. Give example also.

(c) Demonstrate band structure of solids. 5

Unit II

- 3. (a) Explain symmetry chirality and optical activity with example. 8
 - (b) Demonstrate the synthesis and mechanism of asprin.

2

Or

- 4. (a) Show the mechanism of ring opening reaction by taking suitable example. 7
 - (b) Describe all the conformations and potential energy diagram of *n*-hexane. 8

Unit III

- 5. (a) What is critical phenomenon? Derive equations of states of real gases. 8
 - (b) Describe acid-base equilibria in detail. 7

Or

- 6. (a) Give a detailed discussion on temporary and permanent hardness. 8
 - (b) Discuss the use free energy in metallurgy with the help of Ellingham diagram. 7

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P.T.O.

(b)	Gives a discussion on effective nucle	ear
	charge.	2
(c)	Discuss about the diastereomers.	2
(d)	Illustrate addition reaction with examp	le.
		2
(e)	Sumarize the various applications	of
	diffraction.	2
(f)	Discuss potential energy surfaces.	2
(g)	Define cell potential.	2
(h)	What are selection rules for electron	nic
	spectroscopy?	1

(b)	Gives a discussion on effective nuclea	ar
	charge.	2
(c)	Discuss about the diastereomers.	2
(d)	Illustrate addition reaction with example	Э.
	~0	2
(e)	Sumarize the various applications of	of
	diffraction.	2
(f)	Discuss potential energy surfaces.	2
(g)	Define cell potential.	2
(h)	What are selection rules for electronic	ic

450

spectroscopy?